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dissolves in 75.0 g of water in a calorimeter, the temperature drops
from 31.00C to 21.60C. Calculate AH for the process. $\text{KCl(s)} \rightarrow \text{K}^+(\text{aq})$
 $+ \text{Cl}^-(\text{aq})$ even 6. What is the specific heat of gold if the temperature
of a 8.21-g sample of gold is increased by 6.2 co when 6.51 J of heat

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Practice Problems 1. Identify each the following as a formula of an organic or inorganic compound: a. KCl b. C_4H_{10} c. $\text{C}_2\text{H}_6\text{O}$ d. H_2SO_4 e. CaCl_2 f. $\text{C}_3\text{H}_7\text{Cl}$ 2. Identify each of the following as a formula of an organic or inorganic compound: a. $\text{C}_6\text{H}_{12}\text{O}_6$ b. K_3PO_4 c. I_2 d. $\text{C}_2\text{H}_6\text{S}$ e. $\text{C}_{10}\text{H}_{22}$ f. $\text{C}_4\text{H}_9\text{Br}$ 3.

Chapter 12 Introduction to Organic Chemistry: Hydrocarbons

Figure 12.1.4 Dissolution of 1 mol of an Ionic Compound In this case, dissolving 1 mol of $(\text{NH}_4)_2\text{Cr}_2\text{O}_7$ produces a solution that contains 1 mol of $\text{Cr}_2\text{O}_7^{2-}$ ions and 2 mol of NH_4^+ ions. (Water molecules are omitted from a molecular view of the solution for clarity.)

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